Diagramme du zinc

$C = 10^{-6}$ mol.L^{-1}

$E_1 = E^0 + 0{,}03 \log C = -0{,}94$ V

$E_2 = E^0 + 0{,}03\,(28\text{-pKs})\text{-}0{,}06\text{pH} = -0{,}43\text{-}0{,}06$ pH

$E_3 = E^0 + 0{,}03\,(4\text{pKe}-\log \text{Kf}+\log C) = 0{,}27\text{-}0{,}12$ pH

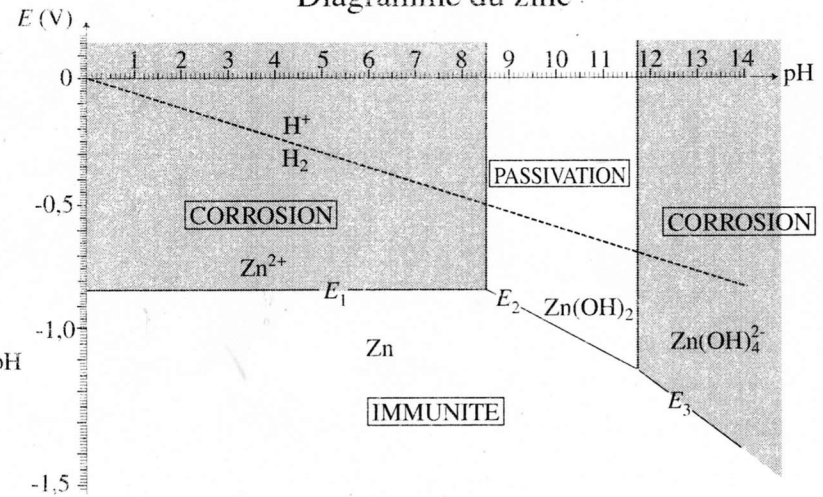

Diagramme du fer

$C = 10^{-6}$ mol.L^{-1}

$E_1 = -0{,}62$ V

$E_2 = 0{,}77$ V

$E_3 = 1{,}07\text{ -}0.18\text{pH}$

$E_4 = -0{,}06\text{ - }0{,}06$ pH