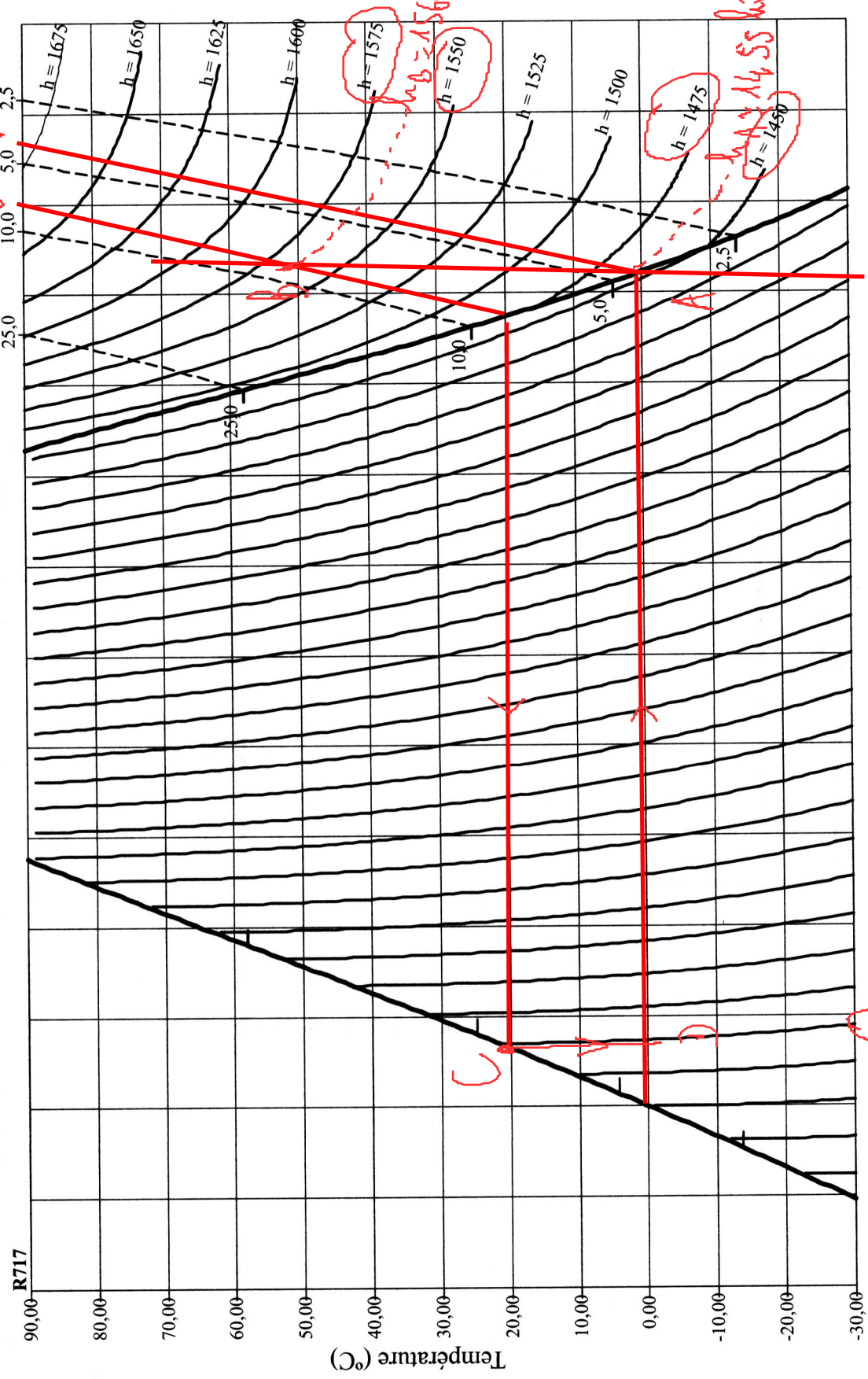


# Diagramme entropique de l'ammoniac

$P_s(20^\circ C) = 8 \text{ bar}$   
 $P_s(0^\circ C) = 4 \text{ bar}$



Temperature (°C)

Entropie massique (J.kg<sup>-1</sup>.K<sup>-1</sup>)

$h_{10} = h_{11} = 300 \text{ J.kg}^{-1}.K^{-1}$

$h_{10} = 1560 \text{ kJ.kg}^{-1}$

$h_{11} = 1455 \text{ kJ.kg}^{-1}$